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Ionic compound - Wikipedia

An acid is a molecule or ion capable of donating a hydron (proton or hydrogen ion $H +$), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or BrÃ¸nsted acids.In the special case of aqueous solutions, proton donors form the hydronium ion $H 3 O +$ and are known as Arrhenius acids.

Acid - Wikipedia

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